E ducation in environmental engineering and science has historically been conducted at the graduate level, and up to the present time has drawn mainly on students with a civil engineering background. In general, education in civil engineering does not prepare a student well in chemistry and biology. Since a knowledge of these sciences is vital to the environmental engineer, the graduate program must be designed to correct this deficiency. In recent years, students from other engineering disciplines and from the natural sciences have been attracted to this field. Some have a deficiency in chemistry and biology similar to that of the civil engineer and need exposure to general concepts of importance.

A current trend in the United States is the introduction of an undergraduate environmental engineering option or degree program within civil engineering departments. These students also require an introduction to important concepts in chemistry and biology.

This book is written to serve as a textbook for a first course in chemistry for environmental engineering and science students with one year of college-level chemistry. Environmental professionals need a wide background in chemistry, and in recognition of this need, *Chemistry for Environmental Engineering and Science* summarizes important aspects from various areas of chemistry. This treatment should help orient the students, aid them in choosing areas for advanced study, and help them develop a better "feel" for what they should expect to gain from further study.

The purpose of this book is twofold: It (1) brings into focus those aspects of chemistry that are particularly valuable for solving environmental problems, and (2) it lays a groundwork of understanding in the area of specialized quantitative analysis, commonly referred to as water and wastewater analysis, that will serve the student as a basis in all the common phases of environmental engineering practice and research.

Substantial changes continue to occur in the emphasis of courses for environmental engineers and scientists. The trend is toward a more fundamental understanding of the chemical phenomena causing changes in the quality of surface and groundwaters, of waters and wastewaters undergoing treatment, and of air. This fundamental understanding of chemistry is absolutely critical as environmental professionals attempt to solve complex problems such as hazardous waste pollution, air pollution from emission of toxic compounds, radioactive waste disposal, ozone depletion, and global climate change.

Chemistry for Environmental Engineering and Science is organized into two parts. Part One is concerned solely with fundamentals of chemistry needed by environmental engineers and scientists. It includes chapters on general chemistry, physical chemistry, equilibrium chemistry, organic chemistry, biochemistry, colloid chemistry, and nuclear chemistry. Each emphasizes environmental applications. In this new edition, the chapters on general and physical chemistry have been updated, and new homework problems have been added. The chapter on equilibrium chemistry has been revised, with many new example and homework problems. The chapter on organic chemistry includes an added emphasis on organic compounds of environmental significance (e.g., chlorinated solvents). Sections are included on the behavior (fate) of organic compounds in the environment and in engineered systems and on the use of structure-activity relationships. The chapter on biochemistry has been updated. We feel that these revisions make the text even more suitable for lecture courses on environmental chemistry principles.

Part Two is concerned with analytical measurements. A new chapter has been added on statistical analysis of analytical data. All analytical procedures are subject to errors. There is a critical need for students to learn how to evaluate the uncertainties such errors present. This chapter discusses basic methods for evaluating and reporting uncertainties in measurements that are essential for analytical chemists, regulatory agencies, and environmental professionals who use analytical data to make important decisions.

The next several chapters contain general information on quantitative, qualitative, and instrumental methods of analysis, useful as background material for the subsequent chapters concerned with water and wastewater analyses of particular interest to environmental engineers and scientists. These chapters are written to stress the basic chemistry of each analysis and show their environmental significance. They should be particularly useful when used with "Standard Methods for the Examination of Water and Wastewater," published jointly by the American Public Health Association, American Water Works Association, and Water Environment Federation, and giving the details for carrying out each analytical determination. The final chapter stresses trace contaminants, many of which are determined analytically with instrumental procedures discussed in earlier chapters. A listing of U.S. Environmental Protection Agency drinking water standards and World Health Organization drinking water quality guidelines for various trace contaminants are also contained in this chapter. Part Two is considered to be most useful as lecture material to accompany a laboratory course in environmental chemistry. Revisions have been made in other chapters to reflect the many changes in "Standard Methods" that have occurred since the last edition of this text.

Problems are included at the end of most chapters to stress fundamentals and increase the usefulness of this book as a classroom text. Example problems throughout the text help increase the students' understanding of the principles outlined. In Part One of the book, where the emphasis is on chemical fundamentals, answers are included after many homework problems, allowing students to evaluate independently their understanding of the principles emphasized.

To meet textbook requirements, brevity has been an important consideration throughout. For those who believe that we have been too brief, we can only beg their indulgence and recommend that they seek further information in standard references on the subject. Important references are listed at the end of each chapter. It is inevitable that we have made errors in producing this textbook. For this we apologize. Hopefully they are not so numerous that they impede the student's ability to learn the material. Fortunately, for this new edition, McGraw-Hill is providing a website where we can list errata that can be readily downloaded with no charge to students and faculty. A solution manual for text problems can also be obtained at this website, but by faculty members only. We hope also to use this website to post more example problems and their solutions. There was a request for such by reviewers, and this is one way that we can provide additional material without expanding the number of pages and costs for the text. The website for this textbook can be found at http://www.mhhe.com/sawyer. We would appreciate hearing from students and faculty when errors are found so that we can enter them in a timely manner on the website. Our e-mail addresses are included in the errata section of the website.

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